	DATED: 00/10/10 TAGE TOF				
PRODUCT SPECIFICATION					
Product Name	Magnesium Silicofluoride Hexahydrate				
Alternative Name	Magnesium Fluorosilicate, Magnesium Hexafluorosilicate				
Specification Reference	MASIF/3 (10/18/0034392)				
PHYSICAL PROPERTIES					
Description	White crystals				
MgSiF ₆ .6H ₂ O	98 % min				
Insoluble Residue in H ₂ O	0.3 % max				
Granulometric analysis type					
>495 microns	15% max				
Apparent density	0.9 - 1.1 kg/l				

NOTES

Exclusion of Liability

Information contained in this publication is accurate to the best of the knowledge and belief of Tennants.

Any information or advice obtained from Tennants otherwise than by means of this publication and whether relating to Tennants materials or other materials, is also given in good faith. However, it remains at all times the responsibility of the customer to ensure that Tennants materials are suitable for the particular purpose intended.

Tennants accepts no liability whatsoever (except as otherwise provided by law) arising out of the use of information supplied, the application, adaptation or processing of the products described herein, the use of other materials in lieu of Tennants materials or the use of Tennants materials in conjunction with such other materials.

Health and Safety

A Material Safety Data Sheet has been issued describing the health, safety and environmental properties of this product, identifying the potential hazards and giving advice on the handling precautions and emergency procedures. This must be consulted fully before handling, storage and use.

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SAFETY DATA SHEET

1. IDENTIFICATION OF THE SUBSTANCE/PREPARATION AND COMPANY

1.1 Product Identifier

Product Name MAGNESIUM SILICOFLUORIDE HEXAHYDRATE

Synonyms Magnesium silicofluoride. Magnesium fluorosilicate. Magnesium

hexafluorosilicate

CAS Number 18972-56-0

REACH Registration Number 01-2119980031-47-XXXX

1.2 Relevant identified uses of the substance or mixture and uses advised against

Identified use(s)

Formulation. Crystallisation. Concrete. Waterproofing additive. Waterproofing additive for cement, concrete hardener, terrazzo flooring polisher, wood preserving, ceramics

Uses advised against:

No data available for uses advised against.

1.3 Details of the supplier of the safety data sheet

Tennants Distribution Limited

Hazelbottom Road

Cheetham

Manchester

ivianches

M8 0GR Tel: 44(0)

Tel: 44(0)161 205 4454 Fax: 44(0)161 203 4298

Email: msds@tennantsdistribution.com

1.4 Emergency telephone number

Tel: 44(0)844 335 0001 (24 hours)

2. HAZARDS IDENTIFICATION

2.1 Classification of the substance or mixture

2.1.1 Regulation 1272/2008 (CLP)

Acute Tox. 3 (Oral) H301

Serious eye damage Cat. 1. Chronic aquatic toxicity Cat.3

2.2 Label elements

2.2.1 According to Regulation (EC) No. 1272/2008 (CLP).

Hazard Pictogram





Signal word(s) Danger.

Hazard statement(s)

H301: Toxic if swallowed.

H318: Causes serious eye damage.

H332: Harmful by inhalation.

H412: Harmful to aquatic organisms, with long-lasting harmful effects.

Precautionary statement(s)

P264: Wash hands thoroughly after handling.

P270: Do not eat, drink or smoke when using this product.

P330: Rinse mouth.

P301+P310: IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician.

P405: Store locked up.

P501: Dispose of contents / container to hazardous or special waste collection point.

Physioco-chemical hazards

The product, heated above 120° C, releases toxic and corrosive fumes.

In contact with strong mineral acids, it produces Hydrofluoric Acid.

Environmental hazards

Fluorides may poison the biota. It is very soluble in water

Human health hazards

Adverse effects are observed by ingestion and acute inhalation.

According to the tests carried out, long-term adverse effects are not expected.

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2.3 Other Hazards

It is not considered a PBT or vPvB

3. COMPOSITION/INFORMATION ON INGREDIENTS

Substances

Chemical family: Inorganic fluoride

Chemical name: Magnesium Hexafluorosilicate

CAS No.: 18972-56-0

4. FIRST AID MEASURES

4.1 Description of first aid measures

Inhalation

Remove the affected person from the danger area. Make him/her as comfortable as possible and protect him/her from cold. If breathing becomes laboured or the patient is cyanotic, give him/her oxygen through a facemask. Go immediately to a doctor.

Skin contact

Wash immediately the skin with plenty of water. Take off contaminated clothes. If irritation persists after washing, seek medical advice.

Eve contact

Wash immediately the eyes with plenty of water for 10 minutes at least, keeping the eyelids open. Seek immediate medical advice.

Ingestion

Seek immediate medical help if possible carrying this SDS or container label. Do not induce vomiting or give drink / eat anything to the injured person.

4.2 Most import symptoms and effects, both acute and delayed

In case of contact it can cause irritation to skin, eyes and respiratory tract. If swallowed, it can cause gastrointestinal irritation and subsequent Fluoride poisoning. The immediate treatment is essential to reduce the severity of the consequences of burns or poisoning. In either case it is always recommended to seek medical attention.

4.3 Indication of any immediate medical attention and special treatment needed

It is strongly recommended the presence of emergency showers and eye baths close to workstations. Because of the singularity of fluorides burns and poisoning, accident assistance and emergency services at local hospitals should be duly informed of the specific and concrete medical treatment required.

5. FIRE FIGHTING MEASURES

The product is non-combustible and non-comburent.

5.1 Extinguishing Media

Use fire-extinguishing media appropriate for surrounding materials.

5.2 Special Fire fighting procedures

Above 120°C, emission of SiF₄ + MgF₂

5.3 Advice for fire-fighters

Remove the containers to a safe area if this operation may be done without danger. When extinguishing fires, breathing apparatuses and full chemical protective clothing should be worn.

6. ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions, protective equipment and emergency procedures

Wear suitable protective clothing (see paragraph 8). Provide good ventilation. Keep away from heat sources. Do not allow the access to release area to people who do not use personal protective equipment.

6.2 Environmental precautions

Recuperate as much product as possible for the productive cycle. Prevent soil, water and drain pollution.

6.3 Methods and material for containment and cleaning up

Mechanically pick-up avoiding dust formation. Introduce the product into clean, dry, sealable and duly marked containers. Move the containers off the spillage area. Then, thoroughly wash the area with plenty of water.

6.4 Reference to other sections

See sections 8 and 13

7. HANDLING AND STORAGE

7.1 Precautions for safe handling

Avoid inhalation, absorption and contact with the product. Carefully handle and open the container avoiding spillage and dust formation. Assure good ventilation. Partially used containers should be hermetically closed after use and returned to the store. The empty containers may contain waste, so they should be handled as if they were full.

7.2 Conditions for safe storage, including any incompatibilities

The store should be situated in a dry and, well- ventilated place, in an area specially adapted to toxic products far from heat sources and separate from incompatible products. Store the product in the hermetically closed original container. Do not store near foodstuffs destined to human or animal nutrition. In Spain, storage must meet the R.D. 379/2001

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(Chemical Products Storage Regulations) if the quantity stored is above 250 Kg.

Plasticized paper bags, plastic bags, plasticized cardboard drums, plastic drums or plastic RIG can be used as packing materials.

7.3 Specific end use(s)

See section 1.2.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

8.1 Control parameters

Limit level for repeated exposures:

VLA-ED: 2.5 mg (F)/m³ – Guide of INSHT Biological limit level: VLB

Biological indicator: Fluorides in urine.

End of working day: 8 mg/l. – Guide of INSHT

Before shift, 4-mg/g creatinine; after shift: 7-mg/g creatinine BAT.

DNEL: Derived No Effect Level

Exposure pattern	Route	Value	Effects	Population
Acute effects (systemic and local)	Inhalation	2.5mg/m ³		workers
Long- term effects (systemic and local)	Inhalation	2.5mg/m ³		workers

PNEC: Predicted No Effect Level

	Value
Freshwater	0.9 mg/l
Seawater	0.9 mg/l
Sediments (freshwater)	No data
Waste treatment plant	51 mg/kg wwt
SOIL	11 mg/kg soil dw

8.2 Exposure controls

Process Conditions

Local vacuuming recommended for keeping dust emissions below the lowest admissible exposure levels. Periodic control of working environment should be done.

Appropriate engineering controls

If handling conditions make dust, it should be necessary to use personal protection equipments. Do not eat, drink or smoke while handling the product. At the end of work, shower or wash.

Before breaks, wash your hands. Change working clothing after handling the product. Change dirty or moistened clothing and wash prior to its reuse. Keep working clothing separated. The shower and washing areas should be separated from changing rooms. Keep the product away from food, beverages and condiments.

Respiratory protection

In accordance with exposure levels, use the respiratory equipment adequate to those levels. The appropriate respiratory equipments –all of them EPI's category 3-, may be self-filtering masks type FFP·, filters with half- face mask type P3, filters with full-face mask type P3, aid-ventilation with helmet or hood type THP3, aid- ventilation with full-face mask type TMP3.

Hand protection

Gloves of rubber or neoprene.

Eye protection

Well-fitted chemical protective goggles type motorist or diver. It is generally known that contact lenses must not be worn when working with chemicals because they may contribute to the severeness of possible damage to the eyes.

Skin protection

In normal conditions, light protective clothing (overall) with long sleeves, and rubber or neoprene boots. Additional, and for emergencies, they should have a particle-proof suit, EPI category 3 type 5, with self- contained breathing equipment.

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Environmental exposure control

Gaseous emission: specific limits as per integrated environmental authorisation.

R.D. 102/2011 – Gaseous emission (out of the manufacturing site):

Fluorides 60µg/m³ (30min) Fluorides 20µg/m³ (1 day)

9. PHYSICAL AND CHEMICAL PROPERTIES

9.1 Information on basic physical and chemical properties

Appearance: White crystalline solid

Odourless

pH: 2-3 dissolved in water at 20° C.

Melting point / Freezing point: No data available

Initial boiling point and boiling range: Approx. 120° C with decomposition

Flash point:

Evaporation rate:

Flammability:

Non-flammable

Non-flammable

Non-flammable

Non-explosive

Vapour pressure: N/A
Vapour density: N/A

Relative density: $0.9 - 1.1 \text{ g.} / \text{cm}^3$

Solubility(ies):

590 g. / 1. at 20° C. Solubility in water: Solubility in other chemical products: No data available Partition coefficient n-Octanol/water: No data available Auto-ignition temperature: Non-flammable Decomposition temperature: 120° C approx. Viscosity: No data available Explosive properties: Non-explosive Non-comburent. Comburent properties:

9.2 Other information

Miscibility: Miscible in water.
Liposolubility: No data available
Conductivity: No data available.

10. STABILITY AND REACTIVITY

10.1 Reactivity

With strong mineral acids, generates HF. Moisture. The product should be kept dry..

10.2 Chemical stability

Stable under normal conditions. It does not decompose when used as per the rules..

10.3 Possibility of hazardous reactions

No data available

10.4 Conditions to avoid

Temperatures above 120°C.

10.5 Incompatible materials

Strong mineral acids: the reaction with strong mineral acids produces HF.

10.6 Hazardous decomposition products

Heating above 120°C, will cause the production of SiF₄ + MgF₂

11. TOXICOLOGICAL INFORMATION

11.1 Information on toxicological effects

Acute toxicity

Ingestion:

Toxic by ingestion. LDL50 = 291 mg / Kg bw

Inhalation:

 $LC50 = 3900 \text{ mg/m}^3$

Skin corrosion/irritation

Not irritant

Serious eye damage/irritation

Causes serious eye damage. Irreversible effects on the eyes

Respiratory or skin sensitisation

Not sensitising

Germ cell mutagenicity

Not considered mutagenic

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Carcinogenicity

There is no evidence of an association between cancer and exposure to inorganic fluorides (IARC).

Reproductive toxicity

Not considered toxic for reproduction

Specific toxicity in certain organs (STOT) - single exposure

In view of the available data, the criteria for classification are not met.

Specific toxicity in certain organs (STOT) – repeated exposure

In view of the available data, the criteria for classification are not met.

Aspiration hazard

Possible irritation of the respiratory tract and possible poisoning by Fluorides. No experimental data available.

Other data

RTECS / NIOSH-ED Toxicity Registration Number: WA 8470000.

12. ECOLOGICAL INFORMATION

12.1 Acute Fish Toxicity

Acute toxicity freshwater fish

LC50 = 100 mg / 1

Long-term toxicity freshwater fish EC10 / LC10 = 4mg / 1

Short-term toxicity for freshwater invertebrates EC50 / LC50 = 100 mg / 1

Long-term toxicity for freshwater invertebrates EC10 / LC10 = 8.9 mg / 1

Toxicity in freshwater algae EC50 / LC50 = 27.4 mg / 1 EC10 / LC10 = 21.6 mg / 1

Toxicity to aquatic microorganisms EC50 / LC50 = 151 mg / 1. EC10 / LC10 = 62.5 mg / 1

Acute toxicity freshwater fish LC50 = 100 mg / 1

Long-term toxicity freshwater fish EC10 / LC10 = 4mg / 1

Short-term toxicity for freshwater invertebrates EC50 / LC50 = 100 mg / 1

Long-term toxicity for freshwater invertebrates EC10 / LC10 = 8.9 mg / 1

Toxicity in freshwater algae EC50 / LC50 = 27.4 mg / 1 EC10 / LC10 = 21.6 mg / 1 EC10

Toxicity to aquatic microorganisms EC50 / LC50 = 151 mg / 1

EC10 / LC10 = 62.5 mg / 1

Toxic effect on fish and plankton, plants and foliage.

Prevent the substance from entering surface water, wastewater and the soil.

12.2 Persistence and degradability

The substance dissociates and hydrolyses in aqueous solution. The overall reaction is very influenced by pH. However, at pH of the environment, the substance is completely hydrolysed, giving rise to the fluoride ion. No biodegradation observed in water.

12.3 Bio accumulative potential

Due to the water-soluble nature of magnesium fluorosilicate, bioaccumulation cannot be evaluated. The substance has a low potential for bioaccumulation due to its rapid hydrolysis.

12.4 Mobility in soil

The product has low mobility in soil. The soil natural alkalinity will slowly dissipate acidity. If pH > 6.5, the soil will strongly bind fluorides. High calcium content will also immobilize fluorides.

12.5 Results of PBT and vPvB assessment

Not considered a PBT or vPvB substance.

12.6 Other adverse effects

No experimental data available

13. DISPOSAL CONSIDERATIONS

Use as much of the product as possible in the production cycle. The product packed in dry sealed vessels should be given to authorised waste agent in order to manage its elimination that will probably be done by means of a physiochemical treatment and later settlement in a controlled chemical waste weir.

Treatment of containers

Remove dust with caution from containers before their disposal or deposit. Once they are almost clean, dispose of them by placing in an authorized dumping place or, in any case, burning them at a legally authorised plant. Non-contaminated packaging will be treated as inert residues or as recyclable material.

Other information

Before any disposal process, take advice of the national, autonomic and local legislation.

An authorised waste manager, or the product manufacturer, may collaborate / advise in said disposal.

14. TRANSPORT INFORMATION

14.1 UN Number	
UN NO.(ADR/RID/ADN)	2853
14.2 UN Proper Shipping Name	
Proper Shipping Name	Magnesium Fluorosilicate

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14.3 Transport Hazard Class		
ADR/DID/ADN Class	6.1 T5	
ADR/DID/ADN Class	Class 6.1: Toxic substances	
ADR Label No.	60	
Transport Labels	6	
14.4 Packing Group		
ADR/DID/AND Packing Group	III	
14.5 Environmental Hazards		
Environmentally Hazardous Substance/ Marine	No	
Pollutant		

14.6 Special Precautions for User

Protect from moisture. Keep away from foodstuff and pharmaceuticals.

14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

MARPOL Annex II: rules to prevent pollution by noxious liquid substances. Not applicable

15. REGULATORY INFORMATION

15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

It is not included in Regulation (EC) 689/2008 on the export and import of dangerous chemicals. Due to its toxicity, it comes into SEVESO category.

15.2 Chemical safety assessment

A Chemical Safety Assessment has been carried out

16. OTHER INFORMATION

This safety data sheet is from Magnesium Hexahydrate fluorosilicate. The magnesium fluorosilicate hexahydrate has the identifier CAS number: 18972-56-0, but there is no EINECS number as identifier. Hexahydrated Magnesium Hexafluorosilicate has no harmonized classification according to Regulation 1272/2008 (CLP Regulation), and the classification indicated here is based on the tests required for the REACH registration. In the list of harmonized classification of the CLP Regulation is listed the Hexafluorosilicato of Magnesium, in its anhydrous form.

Full text of H-Statements referred to under sections 2 and 3

H301: Toxic if swallowed.

H318: Causes serious eye damage.

H332: Harmful by inhalation.

H412: Harmful to aquatic organisms, with long-lasting harmful effects.

Source of key data used to compile the data sheet

Supplier information

Modifications from last revision

The Specification has been updated. The Safety Data Sheet has been updated throughout in accordance with current requirements

Date: 05/10/18

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